PALLADIUM - CATALYZED DIASTEREOSELECTIVE SYNTHESES OF (E)l-TFUMETHYLSILYL-2-A, (E)-l- TRIMETHYLSILYL- 1 -ALKEN-3- YNES, (lE,SE)-I-TRIMETHYLSILYL-1.5~ALKADIEN-3-YNES, (lE,3Z)- AND (lE,3E)-l- TRIMETHYLSILYL- 1.3~ALKADIENES

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Abstract: On the basis of our observation that (E)-1-bromo-1-alkenes undergo preferentially stereospecific Pd-catalyzed cross-couplings with a variety of organometallics, in the presence of the corresponding (Z)-stereoisomers, efficient and convenient cliastereoselective procedures have been developed to prepare nearly stereoisomerically pure (E)-1-trimethylsilyl-2-alkenes (4). (E)-ltrimethylsilyl-1-alken-3-ynes (5), (1E, 5E)-1-trimethylsilyl-1,5-alkadien-3-ynes (6), and (1E,3E)-1**trimethylsilyl-1.3~alkadienes (8) from stereoisomeric mixtures of alkenyl bromides. Compounds 5** have been stereoselectively converted into (1E,3Z)-1-trimethylsilyl-1,3-dienes (7) by selective **hydrometallations. followed by hydrolysis. Some synthetic applications of compounds 5-8 have been also examined.**

Organosilanes containing stereodefmed mono- or polyunsaturated moieties are quite interesting reagents which are tolerant of a wide variety of organic functionality and undergo many synthetically useful transformations with a host of electrophiles². Consequently, efficient **synthetic methods for their preparation from easily accessible substances are desiderable.**

During our previous studies on highly diastereoselective palladium-catalyzed cross-couplings between 1-alkynylzinc chlorides and stereoisomeric mixtures of 1,2-dibromoethene^{3,4} or 1-alkenyl bromides⁵, we have developed efficient procedures for the synthesis of (E)-1,6bis(trimethylsilyl)-hexa-3-en-1,5-diyne (1)^{3,4} and (E)-1-trimethylsilyl-3-alken-1-ynes (2)⁵.

Interestingly, these last compounds can be easily converted into (lZ,3E)-l-trimethylsilyl-1.3-alkadienes (3) by hydroalumination in ether solution, followed by hydrolysis with icecold 3 N NaOH6.7.

In continuation of the studies in the area of the highly diastereoselective palladium-cat-

alyzed carbon-carbon bond forming reactions involving the use of stereoisomeric mixtures of alkenyl bromides⁸, we now wish to report simple and convenient procedures for the synthesis **of (E)-1-trimethylsilyl-2-alkenes (4). (E)-1-trimethylsilyl-1-alken-3-ynes (5). (lE.SE)-ltrimethylsilyl-1,5-alkadien-3-ynes (6). (lE,3Z)-1-trimethylsilyl-1,3-alkadienes (7), and** $(1E,3E)$ -1-trimethylsilyl-1,3-alkadienes $(8)^9$.

These procedures stem from our observation that (E)-1-bromo-1-alkenes undergo preferentially stereospecific palladium-catalyzed cross-couplings with a variety of organometallics, in the presence of the corresponding (Z) -stereoisomers^{3-5,7-10} (eq 1).

$$
n R \rightarrow Br + m R \rightarrow Br + n R^1 M \quad Pd cat \rightarrow n R \rightarrow R \rightarrow Br
$$

[R^1 = aikyl, aryl, 1-alkynyl, PhS; $M = MgX$, Cu, ZnCl, SnBu₃] [1]

Moreover, some synthetic applications of compounds 5-8 will be discussed.

RESULTS AND DISCUSSION

A) Synthesis of (E) -1-trimethylsilyl-2-alkenes $(4)^{11}$

Compounds 4 were diastereoselectively synthesized from stereoisomeric mixtures of alkenyl bromides (9) and trimethylsilyhnethyhnagnesium chloride (10) (eq 2).

Br+m @ d + 0.95 n Me\$KX\$lgCI Pd(PPh,), ether (E) -9 (Z) -9 10 Ff d Bf o%n aSiM% + 0.05n Rd& + m&d+,+ **(E) -4 (E) -9 (Z) -9 121**

Two of the alkenyl bromides employed, i.e. $(E)/(Z)$ -B-bromostyrene $(9a)$ $(R = Ph, R¹ =$ **H)** and (E)/(Z)-2-bromo-2-butene (9c) $(R = R^1 = CH_3)$, were commercially available. A stereoisomeric mixture of 1-bromo-1-octene (9b) $(R = n-C_6H_{13}, R^1 = H)$ was prepared from **1-octene according to a previously reported general procedure for the synthesis of (E)/(Z)-lbromo-I-alkenesaa.**

The following procedure was employed to prepare compounds (E) - $4¹¹$. An ether solution of 0.95 equiv of 10 was added to an ether solution of a diastereoisomeric mixture of an **alkenyl bromide (9). containing n equiv of the (E)-stereoisomer, and 4-7 mol % of** $Pd(PPh₃)₄$. The reaction mixture was maintained at 0 °C for 0.5 h, then stirred at 25-35 °C **for 15-17 h. GLCYMS of this mixture showed the presence of unreacted (Z)-9 and a new compound, 4, in addition to the expected small amount of unreacted (E)-9. Usual work up and purification by MPLC on a silica gel column, followed by fractional distillation, allowed to obtain the desired (E)-1-trimethylsilyl-2-alkene, having 97-99 % stereoisomeric purity, in good isolated yield (70-80 %). Since there was no coupling in the absence of the palladium catalyst, the reaction was catalytic in palladium.**

Interestingly, either 1-bromo-1-alkenes such as $9a$ (R = Ph, R¹ = H) and 9b (R = n- C_6H_{13} , $R^1 = H$), or a 1,2-dialkyl substituted vinyl bromide, i.e. 9c (R = $R^1 = CH_3$), under**went the highly diastereoselective cross-coupling with 10 (Table 1).**

Compound 4a ($R = Ph$, $R^1 = H$) had been previously reported^{12,13}. Evidence for the stereochemistry of 4b ($R = n - C_6H_{13}$, $R^1 = H$) was obtained from its ¹H NMR spectrum which exhibited a typical trans-olefinic coupling constant $(J = 15.1 \text{ Hz})$. On the other hand, ¹H NMR analysis of compound 4c ($R = R¹ = CH₃$), also using nuclear Overhauser enhance**ment (NOE) experiments, did not allow to confirm the (E)-configuration of its carbon-carbon** double bond¹⁴. However, on the basis either of the stereochemical results obtained in the **preparation of** *4a* **and 4b, or of the fact that the expected amount of (Z)-2-bromo-2-butene** (9c) was present in the final reaction mixture derived from the diastereoselective crosscoupling between 10 and (E)/(Z)- $9c$, the (E)-configuration was attributed to compound $4c$.

TABLE 1

Diastereoselective synthesis of (E)-1-trimethylsilyl-2-alkenes (4)^{a)}

a) All reactions were carried out using 0.95 equiv of 10 and 4-7 mol % of Pd(PPh₃)₄

Owing to the high yields and stereoisomeric purities, as well as the simplicity and the possibility of large scale preparations, this diastereoselective procedure was more convenient than those based either on the stereospecific Ni- or Pd-catalyzed cross-coupling between *10* **and** (E)-alkenyl iodides^{11b,c}, or on the coupling between alkyllithium compounds and (E)-1-iodo-2-methyl-3-trimethylsilyl-1-propene^{11b}. In fact, the usefulness of these stereospecific proce-

dures is limited by the fact they require the use of stereoisomerically pure alkenyl iodides, which are synthesized by several manipulations involving the use of expensive reagents¹⁵.

B) Synthesis of (E) -1-trimethvlsilvl-1-alken-3-ynes $(5)^{16}$ and $(1E.5E)$ -1trimethylsilyl-1.5-alkadien-3-ynes $(6)^{17}$

The following typical procedure was developed to prepare (E)-l-trimethylsilyl-l-alken-3 ynes (5) from a commercially available diastereoisomeric mixture of 2-(bromovinyl)trimethylsilane (II). Treatment of a THF solution of (E)/(Z)-11, containing 1 equiv of the (E) stereoisomer, with 3-4 mol % of Pd(PPh₃)₄ and 0.95 equiv of a THF solution of a 1alkynylzinc chloride $(12a-d)$ ($R = n-C₅H₁₁$, Ph, Me₃Si, 2-thienyl) at -20+0 °C for 2-15 h afforded the desired compound $5a-d$ ($R = n-C_5H_{11}$, Ph. Me₃Si, 2-thienyl), having stereo**isomeric purity higher than 99.5 %, in good isolated yield (77-88 %) (eq 3).**

Interestingly enough. the diastereoselective alkynylation of *(E)/(Z)-11 was* **also accom**plished by treatment of a DMF solution of II , containing 1 equiv of $(E)-II$, with 0.95 equiv of a 1-alkynyltrimethylstannane, in the presence of Pd(PPh₃)₄. However, the isolated yield **was low. Some typical results are reported in Table 2 (entries l-5).**

Of particular significance is that this same procedure. which is more convenient than that previously employed to prepare (E)-1-trimethylsilyl-1-alken-3-ynes such as $5b$ (R = Ph)¹⁶, **when (E)-3-en-1-ynylzinc chlorides were used as alkynylating reagents, allowed to prepare difficult to obtain (lE,5E)-l-trimethylsilyl-1.5-alkadienes (6). Thus, (lE,5E)-l-trimethyl** $silyl-1,5-nonadien-3-yne$ (6a) $(R = (E)-1-pentenyl; R¹ = H)$, having 99 % stereoisomeric **purity, was prepared in 65 % isolated yield starting from** *(E)/(Z)-11 and* (E)3-hepten-l**ynylzinc chloride (I2f) (entry 6. Table 2). Analogously, (E)- 1-trimethylsilyl-4-(l-cyclo**hexenyl)-1-buten-3-yne (6b) $(R, R^1 = -(CH_2)_4)$ was obtained in 81 % yield from (1**cyclohexenyljethynylzinc chloride** *(12g)* **(entry 7, Table 2).**

TABLE 2

Diastereoselective synthesis of (E)-1-trimethylsilyl-1-en-3-ynes (5) and (1E,5E)-1-trimethylsilyl-1,5-dien-3-ynes (6)^{a)}

--N - SiMoa $R-\equiv -ZnCl$ $6a$ R = $^{n}C_3H_7 \rightarrow R^1 = H$ $12f$ R = $^{n}C_3H_7$ 6b R, $R^1 = -(CH_2)_4$ 12g

However, this procedure did not appear suitable to prepare (E)-1-trimethylsilyl-1-buten3 yne (Sf, in good yield by reaction of (E)/(Z)-ZZ with a THF solution of ethynylzinc chloride. In fact, the low boiling point of Sf caused difficulties in isolation of this pure compound from the THF solution of the reaction mixture. Therefore, a different diastereoselective synthesis of *Sf was* **devised (Scheme 1).**

SCHEME 1

Thus, according to a previously reported diastereoselective method to prepare (E)-o-

hydroxy-1,3-enynes^{8a} having high stereoisomeric purity, a stereoisomeric mixture of 11, **containing 1 equiv of (E)-II, was reacted at room temperature with 0.95 equiv of commercially available 2-methyl-3-butyn-2-01 (26). in the presence of catalytic amounts of** Pd(PPh₃)₄ and CuI. The reaction, which was carried out under phase-transfer conditions, em**ploying benzene as organic solvent, a large excess of 3.5 N NaOH as base, and benzyltriethylarnmonium chloride as phase-transfer agent, gave pure (E)-6-trimethylsilyl-2-methyl-5-hexen-3-yn-2-01 (17) in 84 % yield. Reaction of a tetrahydronaphtalene solution of 17 with an equimolar amount of solid NaOH at 125 "C for 1 h gave compound Sfin 68 % isolated yield. Compound** *Sf* **represents an interesting C4 building block not yet reported in the literature.**

Taking into account that the synthesis of *Sf* **did not require the use of costly reagents and it was accomplished in a satisfactory overall yield (57 %), the diastereoselective synthesis** reported in Scheme 1 appears more convenient than that based on the preparation of $5c$ ($R =$ SiMe₃) and the subsequent selective desilylation of this compound by reaction with KOH or K₂CO₃ in methanol or with KF·2H₂O in DMF¹⁸.

C) Synthesis of (1E.3Z)- and (1E.3E)-1-trimethylsilyl-1.3-alkadienes. $(7)^{19}$ and (8)²⁰, respectively

Although several methods have been reported for the synthesis of (lE,32)-l-trimethylsilyl-1,3-dienes (7)'9, many of these methods are either limited in scope or require precursors which are not readily available. Thus, taking into account that a variety of nearly stereoisomerically pure (E)-1-trimethylsilyl-1-en-3-ynes (5) could be easily synthesized in one step and high yields, we tested the possibility to reduce (Z)-stereoselectively these enynes using some procedures previously employed to prepare (Z)- 1-alkenylsilanes via hydroalumination 21 , hydromagnesiation²², or hydroboration²³ of the corresponding 1-alkynylsilanes (eq 4).

As shown in Table 3. different methods had to be employed to convert selectively alkyl or silyl and aryl substituted (E)-1-trimethylsilyl-1-en-3-ynes (5) into the corresponding (lE, 3Z)-dienes (7) in satisfactory yields.

Thus, (1E,3Z)-1-trimethylsilyl-1,3-nonadiene (7*a*) ($R = n-C₅H₁₁$) was obtained in 71 % **yield and 99.3 % stereoisomeric purity by application of a hydroboration-protonolysis sequence (entry 3. Table 3). Other procedures based on the reaction of Sa with diisobutylaluminium hydride in hexane, followed by treatment with cold 3 N NaOH (entry 1). or on** the hydromagnesiation of $5a$ with isobutylmagnesium bromide, in the presence of a catalytic amount of Cp₂TiCl₂, followed by hydrolysis of the resulting Grignard reagent (entry 2), were **plagued with low yields as well as stemoisomeric purity lower than that obtained in entry 3.**

Entry	Product : 7			Yield	E/Z
	Compound	R	Procedure	(%)	ratio ⁸⁾
	æ	nC ₅ H ₁₁	i) i-Bu ₂ All-I, hexane, 4 h at 25 °C, 25 hat 45 °C; ii) 3 N NaOH, 0 °C	38b,c	94.2/5.8
2	72	n CgH n_1	i) i-BuMgBr, Cp ₂ TiCl ₂ , ether, 20h; ii) HO,0°C	ვებ	425/57.5
з	æ	nC ₅ H ₁₁	i) Sia ₂ BH, THF, 3h, 20°C; ii) AcOH, 7h, 60℃; iii) H2O2, OH, 3.5h, 40℃	71 ^d	95.2 / 4.8 $(99.3/0.7)^0$
4	70	CeHs	i) i-Bu ₂ AlH, hexane, 24 h, 50 °C; ii) 3 N NaCH 0°C	19 ^{d,e}	84.3/16.7
5	70	C_aH s	i) i-Bu ₂ AlH, ether, 30 h, 35 °C; I) 3NNaOH0°C	38 ^{d,c}	96.2 / 3.8
6	70	C_aH	i) Sia ₂ BH, THF, 3h, 20°C; ii) AcOH, 7h, 70℃; iii) H2O2, OH, 35h, 40℃	59 ^d	$(91.9/8.1)^9$
	$\boldsymbol{\pi}$	Me₃Si	i) i-Bu ₂ AlH, ether, 35 °C, 7h; ii) 3 N NaCH 0°C	87 ^d	99.4 / 0.6 $(99.4/0.6)^9$

TABLE 3

Synthesis of (1E,3Z)-1-trimethylsilyl-1,3-dienes (7) from (E)-1-trimethylsilyl**l**-en-3-ynes (5)

a) **Evaluated by GLC/MS and 'H NMR analyses; b) Evaluated by GLC analysis of the crude** reaction mixture; c) The reaction mixture contained only the desired (1E,3Z)-silylated butadiene **7** and unreacted 5; d) Isolated yield; e) This compound was contaminated by ca. 20 % of the **corresponding (E)-monounsaturated silyl alkenes: f) E/Z ratio for tbe crude reaction mixture; g) lQZ ratio for the isolated product.**

On the other hand, hydroalumination of compound δc ($R = Me₃Si$) in ether, followed by **hydrolysis of the derived dienylalane with ice-cold 3 N NaOH, was used to prepare (lE,32)-** 1,4-bis(trimethylsilyl)-1,3-butadiene (7c) in 87 % yield and 99.4 % stereoisomeric purity **(entry 7). This same method, in spite of the modest conversion, appeared suitable to prepare (lE,32)-1-trimethylsilyl-4-phenyl-1,3-butadiene (7b) having 96.2 % stereoisomeric purity (entry 5).**

The synthetic method we developed to prepare 4-alkyl substituted (1E,3E)-1-trimethylsilyl-**1,3-dienes (8)20 was straightforward and easily amenable to large scale preparations. In fact, these compounds, having stereoisomeric purity higher than 99 %, were synthesized by cross-coupling between a stereoisomeric mixture of 2-(bromovinyl)trimethylsilane (II) and** easily available (E)-1-alkenylalanes (18), in the presence of catalytic amounts of $Pd(PPh₃)₄$ **(eq 5).**

Thus, when a THF solution of a stereoisomeric mixture of II, containing 1 equiv of (E)- II, was allowed to react at 40 $^{\circ}$ C for 16 h with a hexane solution of (E) -1-heptenyl diisobutylalane (18a), in the presence of 3 mol % of Pd(PPh₃)₄, (1E,3E)-1-trimethylsilyl-**1,3-nonadiene (8a) was obtained in 74.7 % yield. Analogously, (lE,3E)-l-trimethylsilyl-5,5-dimethyl-1.3-hexadiene (8b) was prepared in 84 % yield starting from (E)-3,3-dimethyl-** 1-butenyl diisobutylalane (18b). Compounds 18a and 18b were prepared by reaction of a **hexane solution of diisobutylalane with 1-heptyne and 3,3-dimethyl-1-propyne. respectively, at 50 "C for 6 h.**

n B' SiMe₃ + m B' SiMe₃ + n R A(iBu)₂
$$
\frac{Pd(PPh_3)_4}{THF, hexane}
$$

\n(E) -11 (Z) -11 18a R = n-C₅H₁₁ 40 °C, 16 h
\n18b R = tBu
\n $6a$ R = n-C₅H₁₁ (Z) -11
\n8b R = tBu (Z) -11 [5]

It must be noted that the usefulness of this simple and efficient diastereoselective procedure, which represents a new example of palladium-catalyzed highly diastemoselective cross-couplings involving the use of stereoisomeric mixtures of alkenyl bromides, is limited to the preparation of 4-alkyl substituted (lE,3E)-1-trimethylsilyl-1,3-dienes. In fact, while (E)-Zalkylethenyl diisobutylalanes such as Z8a and *Z8b are* **easily and selectively prepared by hydroalumination of the corresponding 1-alkynes, hydroalumination of (hetero)arylacet**ylenes gives rise to mixtures of (E) -1-alkenyl and 1-alkynylalanes²⁴. On this basis, no **attempt was performed to use the above reported palladium-catalyzed alkenylation procedure for the synthesis of 4-(hetero)aryl substituted (1E,3E)-1-trimethylsilyl-l,3-dienes.**

D) Utility of compounds $5 - 8$ as synthetic intermediates

With efficient stereoselective methods for the preparation of polyunsaturated organosilanes available, we turned our attention to some their synthetic applications. First of all, some electrophilic substitution reactions were examined. Reaction of (1E,3E)-8a with 2-thiophene carbonyl chloride (19), in the presence of AlCl₃, take place readily at 0° C to give stereospecifically, upon hydrolysis, $(1E,3E)-1-(2-thenoyl)-1,3-nonadiene (20)$ in 37 % yield (eq 6).

Interestingly enough, treatment of (lE,3E)-8a with dichloromethyl methyl ether (21) at

-78 T. in the **presence of TiCb, upon hydrolysis at low temperature, gave pure (2E,4E)-2,4 decadienal (22) in 40 % yield (eq 7). Compound 22 is a flavouring chemical present in** peanut, cottonseed and soybean oils²⁵. Analogously, reaction of (1E,3Z)-7a with 21 at -78 **OC, in the presence of TiC4. produced stereospecifically (2E,4Z)-2.4~decadienal (23) in 45.1 % yield (eq 8). Compound 23 is one of the volatile constituents of black tea26.**

The l-chloromethoxymethylation of (E)-1-trimethylsilyl-1-en-3-ynes (5). with replacement of the silyl group, appeared also interesting. In particular, the l-chloromethoxymethylation of (E)-1-trimethylsilyl-4-(2-thienyl)-1-buten-3-yne (Sa) could provide a convenient access to (E)-S-(2-thienyl)-2-penten-4-ynal (24). which is an acetylenic thiophene isolated from *Anthemis saguramicd7.*

However, treatment of $5d$ with $2I$ in dichloromethane at -78 °C, in the presence of TiCl₄, **produced a relevant amount of a tarry material GLCYMS analysis of the reaction mixture, after hydrolysis, using an authentic sample of 24 as standard, showed the absence of this aldehyde. The authentic sample of 24, which we also needed for biological studies, was** prepared according to the reaction sequence reported in Scheme 2.

SCHEME 2

Thus, according to an improved Sonagashira procedure²⁸, 2-ethynylthiophene $(25)^{29}$ was coupled with (E) -3-iodo-2-propen-1-ol $(26)^{18c}$, in the presence of catalytic amounts of **Pd(PPh3)4 and CuI. Compound 27 was obtained in 54.3 % yield. This pure alcohol was oxidized using manganese dioxide30 to provide the desired aldehyde in 53.7 % yield.**

Afterwards, a simple application of (lE,5E)-1-trimethylsilyl-l.5-dien-3-ynes (6) to the synthesis of 4-methylcyclopenten-2-ones was examined. The Nazarov cyclization³¹ of (E)-1**trimethylsilyl-4-(1-cyclohexenyl)-1-buten-3-yne (6b) by reaction with a mixture of acetic** acid and sulfuric acid at 60 °C for 40 min gave a mixture containing two regioisomeric ketones, 28a and 29, which were separated by MPLC on a silica gel column (eq 9).

Compound 29, which was recovered in 17.5 % yield from the first eluted fractions, **exhibited a complex IH NMR spectrum. However, the structure of this ketone, which corresponded to 3-methyl-5,6,7,7a-tetrahydroindan-l-one, was elucidated using 2D NMR spec**troscopy. In fact, a 2D¹H NMR correlation $(COSY)^{3}$ spectrum helped to define two iso**lated spin systems in the molecule. These systems indicated the following moieties:**

$$
\begin{array}{cccc}\n1 & 2 & 3 & 3a \\
C - C + I_2 - C + - C & \text{and} & \begin{array}{c}\n1 - C - C + I_2 - C + I_2 - C + I_2 \\
1 - I_3 & 3a & 4\n\end{array} & 6 & 7\n\end{array}
$$

This structure determination was confirmed by single frequency decoupling experiments. The less easily eluted ketone, i.e. 3-methyl-4,5,6,7,-tetrahydroindan-l-one *(28a).* **was obtained in 65 % yield. Its physical and spectroscopic properties were in agreement with those previously reported33.**

It must be noted that few years ago an analogous Nazarov cyclization of (E)-1-trimethylsilyl-4-[(E)-1-dodecenyl]-2-buten-3-yne (6c) allowed to prepare cyclopentenone 28b in 64 **% yieldIT. However, no mention on the formation of an isomeric cyclopentenone was** reported in this communication¹⁷.

Finally, in order to evaluate the possibility to use polyunsaturated organosilanes as organometallic reagents in palladium-catalyzed carbon-carbon bond forming reactions, we examined the reactivity of compounds *Sa. 7a.* **and 8a with some (E)-1-iodo-l-alkenes and aryl iodides. in the presence of equimolar amounts of tris(diethylamino)sulfonium difluorotrimethylsilicate (TASF) and catalytic amounts of q3-allylpalladium chloride. In fact, Hatanaka** and Hiyama³⁴ have recently reported that vinyl-, ethynyl-, and allylsilanes react with aryl, **vinyl and ally1 halides, in the presence of TASF and q3-allylpalladium chloride, to give stereospecifically the corresponding coupled products in synthetically useful yields (eq 10).**

$$
R^{1}X + R-SiM\Theta_{3} \xrightarrow{\text{TASE}} R-R^{1}
$$
\n
$$
R^{1} = \text{aryl, alkenyl, allyl}
$$
\n
$$
R = \text{vinyl, ethynyl, allyl}
$$
\n
$$
R = \text{vinyl, ethynyl, allyl}
$$
\n
$$
R = \text{vinyl, ethynyl, allyl}
$$

However, it was found that the trimethylsilyl substituted organosilanes 5*a*, 7*a*, and 8*a* did not undergo the coupling reactions under the conditions reported in literature³⁴. More **recently, these unsuccessful results have been confirmed by Hatanaka and Hiyama35 who found that the scope of the reactions reported in eq 10 is severely limited mainly due to low reactivity of trimethylsilyl substituted organosilanes compared with other organometallics.**

Since introduction of a fluorine atom into the silicon substituent extremely accelerates the palladium-catalyzed reactions carried out in the presence of TASF or TBAB35, we are now developing synthetically useful methods to prepare diastereoselectively mono- and polyunsaturated stereodefined dimethylfluorosilyl substituted organosilanes.

EXPERIMENTAL

All boiling and melting points are uncorrected. IR spectra were determined on a Perkin-Elmer 283B spectrophotometer. IH NMR spectra were recorded on a Varian Gemini 200. a Varian VXR 300, or a Brucker 360 spectrometer, using TMS as an internal standard. Mass spectra were recorded on a VG 70-70E mass specaometer interfaced with a Dani 3800 gas-chromatograph. *GLC analyses were performed* **on a Dani 6500 gas-chromatograph equipped with a Perkin Elmer LCI-100 integrator. Three types of capillary columns were** used: a PEG-Permaphase bonded FSOT column (25 m x 0.23 mm i.d.), a SRL-300 bonded FSOT column (30 **m x 0.25 mm i.d.). and a SRL-150 bonded FSOT column (30 m x 0.25 mm i.d.). Purifications by MPLC were performed on a Jobin-Yvon Chromatospac Prep 10 liquid chromatograph using a Knauer differential refractometer as detector. All reactions of air and water sensitive materials were performed in flame-dried glassware under an atmosphere of nitrogen or argon. All solvents were dried by conventional methods, freshly distilled and deareated. Air and water sensitive solutions were transferred with hypodermic syringes or double-ended needles.**

Starting materials - Tetrakis(triphenylphosphine)palladium was prepared according to the literature³⁶. Titanocene dichloride, (E)/(Z)-B-bromostyrene (9a), (E)/(Z)-2-bromo-2-butene (9c), (E)/(Z)-2-(bromo**vinyl)trimethylsilane (II). 2-methyl-3-butyn-2-01 (16). and I-ethynylcyclohexene were commercially available. The following compounds were prepared according to literature procedures: (E)/(Z)-l-bromo-l**octene (9b)^{8a}, (E)-3-iodo-2-propen-1-ol (26)^{18c}, 2-ethynylthiophene (25)²⁹. 1-Hexynyltrimethyl**stannane (128) (b-p. 80.5 "C/12 torr) was prepared by reaction of an ether solution of 1-hexynyllithium** with an equimolar amount of trimethyl chlorostannane at $0 \text{ }^{\circ}C^{15}$.

General procedure for the synthesis of (E)-1-trimethylsilyl-2-alkenes (4) - In a typical procedure, to a suspension of $Pd(PPh₃)₄$ (6.0 g, 5.2 mmol) in THF (150 ml) was added a stereoisomeric **mixture of an alkenyl bromide (9) (0.12 mol) containing 1 equiv of the (E)-stereoisomer. The resulting** mixture, which was stirred at 20 °C, became homogeneous within 10 min.

A solution of trimethylsilylmethylmagnesium chloride (10) (0.1 mol) in THF (130 ml) was immediately added to this mixture maintained at 0 "C. The resulting reaction mixture, which was monitored by GLC until ca. 95 % of (E)-9 was consumed, was stirred at the temperature reported in Table 1. Ice-cold waler was added and the mixture was extracted with ether. The extracts. which were analyzed by GLC, were filtered through Celite. dried and concentrated under reduced pressure. The residue was diluted with hexane

and filtered. The filtrate was concentrated under reduced pressure and the residue was purified by MPLC on a silica gel column, using hexane as eluant. followed by fractional distillation on a Fisher column (Spaltrohr-system). Yields and stereoisomeric purities of (E)-1-trimethylsilyl-2-alkenes (4) so obtained am reported in Table 1. *Some* **physical and spectroscopic properties of** *compounds 4 are* **reported below.**

(E)-1-Trimethylsilyl-3-phenyl-2-propene (4a) - B.p. 122-3 °C/15 Torr. ¹H NMR (CCld): ∂ 7.35-**6.90 (SH. m, H2' and H6'). 6.28-6.00 (2H. m. H2 and H3). 1.76-1.50 (2H. m. Hl), 0.05 ppm (9H. s. CH3)**

IR (film): 3080, 3060, 3020, 2900. 2890, 2790. 1940. 1870, 1800, 1740. 1640. 1595, 1490. 1445. 1400, 1300. 1280, 1245. 1205. 1145, 1070, 1020. 1000. 970. 955. 900. 855. 840. 830, 780. 750. 735, and 68s *cm*⁻¹. These physical and spectroscopic properties matched with those previously reported^{12,13}.

(E)-I-Trimethylsilyl-2-nonene (4b) - **B.p. 95-96 "C/11 Torr. 1H NMR (CDC13): i3 5.38 (1H. dt, J E 15.1 and 7.4 Hz. H2). 5.23 (lH, dt. J = 15.1 and 6.0 Hz. H3). 1.97 (2H. br q. J = 6.0 Hz. H4). 1.40 (2H. d. Hl), 1.45-1.16 (8H. br, HS. H6. H7. and H8), 0.88 (3H. t, J = 6.4 Hz. H9). -0.02 (9H. s, CH3). IR (film): 3005 2950. 2920. 2870. 2850. 1660. 1465. 1435. 1400. 1375. 1290. 1245, 1150. 1090. 1055.** 1025, 955, 840, 740, 715, and 685 cm⁻¹. MS (m/e): 199 (M+1, 3), 198 (M, 15), 184 (3), 183 (13), 155 **(3). 127(3). 97 (7). 75 (8). 74 (20). 73 (100). 69 (5). 67 (5). 59 (19). SS (8). 54 (11). 45 (12). 43 (11).** Found : C, 72.61; H, 13.08. Calc. for C₁₂H₂₆Si : C, 72.64; H, 13.21.

(E)-I-TrimethyIsilyI-2-methyl-2-butene (4~) - **B.p. 84-8s *C/150 Torr. IH NMR (CDC13): 8 5.01** $(1H, br q, J = 6.0 Hz, H3), 1.68-1.42$ (8H, br m, allylic protons), 0.02 ppm (9H, s, Si(CH₃)₃). IR (film): **3045. 3020. 295s. 2910. 2860. 1660. 1440. 1410. 1380. 1245. 1210. 1160. 1000. 850. 840. 745. 685. and 650 cm-l. MS (m/e): 142 (hf. 10). 127 (3). 99 (5). 8S (4). 75 (4). 74 (8). 73 (100). 59 (12). 45 (10). 43 (8). Found : C. 67.47; H, 12.68. Calc. for CgHIaSi : C. 67.51; H. 12.75.**

*General procedure for the synthesis of (E)-l -trimethylsilyl-J -alken-3-ynes (5) and (IE,SE)-1-trimethylsilyl-1,S-alkadien-3-ynes (6) - In a typical experiment, a freshly distilled 1***alkyne (50 mmol) was slowly added to a solution of ethylmagnesium bromide (SO mmol) in THF (70 ml) maintained under stirring at 65 "C. After the addition was complete, heating was continued for 1 h. The** THF solution was added to a solution of $ZnCl_2$ (6.81 g, 50 mmol) in THF (50 ml) cooled at -20 °C and the resulting mixture was stirred for 0.5 h. A solution of $Pd(PPh₃)₄$ (1.73 g, 1.5 mmol) and (E)/(Z)-2-**(bromovinyl)trimethylsilane (II) (10.6g. 59 mmol) (E/Z = 89/l 1) in THF (45 ml). which was prepared immediately prior to use, was added and the resulting mixture was stirred at the temperature reported in Table 2. The reaction mixture, which was monitored by GLC until ca. 95 % of (E)-11 was consumed, was** stirred for the required lenght of time (Table 2). Then, it was quenched by addition of a saturated NH_dCl **solution, diluted with water, and extracted with ether (4 x 60 ml). The combined extracts were washed with** water, filtered through Celite, dried and concentrated in vacuo. The residue was diluted with hexane and filtered. The filtrate was concentrated in vacuo and purified by fractional distillation. Yields and stereoisomeric purities of (E)-1-trimethylsilyl-1-alken-3-ynes (5) and (1E,5E)-1-trimethylsilyl-1,5**alkadien-3-ynes (6) so prepared are reported in Table 2. Some physical and spectroscopic properties of compounds 5 and 6. as well as the starting materials** used **in their preparation, are reported below,**

(E)-l-Trimethyisilyl-I-nonen-3-yne (Sa) - **This compound, which was prepared in 78 % yield from** 1-heptynylzinc chloride ($I2a$) and (E)/(Z)- II , had: b.p. 97 °C/3 Torr. ¹H NMR (CDCl₃) : 3 6.29 (1H, **d, I = 20.1 Hz, Hl), 5.94 (lH, dt, J - 20.1 and 3.0 Hz. HZ). 2.27 (2H. td. J - 6.2 and 2.2 Hz. HS). 1.56- 1.44 (2X. br m, H6). 1.42-1.22 (4H. br m. H7 and HE). 0.87 (3H. t, H9). 0.07 ppm (9H. s. SiiCH3)3). MS (m&: 194 (M. 9). 181 (18). 179 (100). 139 (10). 138 (46). 123 (54). 109 (16). 95 (IS), 85 (14). 83** (24), 73 (58), 59 (94), 45 (16). IR (film): 3000, 2960, 2930, 2900, 2890, 2880, 2200, 1575, 1470, 1460, 1430, 1380, 1330, 1260, 1250, 1210, 1050, 980, 860, 845, 735, 720, and 685 cm⁻¹. Found : C, 73.92; H, 11.51. Calc. for C₁₂H₂₂Si : C, 74.14; H, 11.40.

(E)-I-Trimcthylsilyl4-phenyl-1 -buten-3-yne (Sb) - Thir **compound which was prepared in. 88 %** yield from phenylethynylzinc chloride $(I2b)$ and $(E)/(Z)-II$, had: b.p. 85-86 °C/0.2 Torr. Lit¹⁶ b.p. 140 **°C/15 Torr.** ¹H NMR (CC1₄): ∂ 7.39-7.16 (5H, m, C₆H₅), 6.42 (1H, d, J = 19.2 Hz, H1), 6.07 (1H, d, J = **19.2 Hz.** *H2).* **0.10 ppm (9H. I, Si(CH3)3). MS (m/e): 200 (M. 34). 185 (98). 181 (18). 160 (17). 159 (100). 145 (11). 129 (8.5). 73 (8). 59 (18). 43 (13). IR (film): 3090. 3070. 3040. 3000. 2960, 2200. 1600. 1575. 1490. 1445. 1265. 1250. 1210. 1070. 1030,lOOS. 975.910, 860. 840.750. 730, 714 and 690 cm-t.**

(E)-1,4-Bis(trimethylsilyl)-1-buten-3-yne (5c) - This compound, which was prepared in 77 % yield from trimethylsilylethynylzinc chloride $(I2c)$ and $(E)/(Z)-II$, had: b.p. 88 °C/16 Torr. ¹H NMR **(CDC13): a 6.51 (lH, d, J = 19.5 Hz, HI). 5.97 (1H. d. J I 19.5 Ha, H2). 0.19 and 0.08 ppm (18H. 2s.** Si(CH₃)₃). MS (m/e): 196 (M, 15), 182 (20), 181 (100), 155 (20), 123 (15), 119 (12), 97 (10), 83 (18), **72 (82). 70 (52). 45 (10). IR (film): 3000. 2960. 2900. 2150. 2080. 1570. 1410. 1250. 1050. 975, 840,** 755, 740, 735, 690, and 635 cm⁻¹. Found : C, 61.55; H, 9.44. Calc. for C₁₀H₂₀Si₂ : C, 61.14; H, 10.26.

(E)-I-Trimethylsilyl4-(2-thienyl)-I-buten3-yne (Sd) - This **wmpound, which was prepared in 87 % yield from 2-thienylethynylxinc chloride (Z2d) and (E)/(Z)-ZZ, had: b.p. 82 "C/O.02 Torr. lH NMR (CDC13): a 7.24 (1H. dd. HS;'). 7.18 (lH, dd. H3'). 6.96 (1H. dd. H4'). 6.53 (1H. d, J = 19.3 Hz, Hl or**

IR (film): **3015. 3075. 2990. 2955. 2890. 2180. 1570. 1420. 1400. 1295, 1260. 1250, 1205. 1195. 1075. 1040. 970. 860. 845. 835. 770. 730. 720.690, 655. and 610 cm -1. MS (m/e): 207 (M+l. 8). 206 (hf. 45). 193 (7). 192 (14). 191 (84). 167 (9). 166 (15). 165 (100). 151 (7). 73 (18). 59 (13). 45 (7). 43 (10). Found: C, 64.02; H. 6.84. Calc. for C11H14SSi** : C. **64.25; H. 7.00.**

(I&SE)-I-Trimethylsilyl-l,S-nonadien3-yne (6a) - Tbis **wmpound was obtained in 65 % yield** from (E)-3-hepten-1-ynylzinc chloride (*I2f*) and (E)/(Z)-*I1*. Compound *12f* was prepared from (E)-3hepten-1-yne^{18c}, according to the procedure described for the preparation of *12a-d*. Compound 6a had: **b.p. 95-96 'C/2 Torr. 'H NMR (CDC13): 3 6.39 (1H. d. J - 19.2 Hz. Hl), 6.15 (lH, Qt. J = 15.9 and 7.2** Hz, H6). 6.04 (1H, dd. J = 19.2 and 1.8 Hz, H2). 5.60 (1H. br dd. J =15.9 and 1.8 Hz, H5). 2.10 (2H, dq. J . - **7.2 and 1.3 Hz. H7). 1.43 (W. m. J - 7.3 Hz. H8). 0.91 (3H. t. J * 7.3 Hz. H9). and 0.08 ppm (9H. a. Si(CH3)3). IR Wxn): 3020. 2990. 2954 2920. 2890. 2865. 2830. 2175. 1565. 1455, 1430. 1375, 1335, 1330. 1245. 120% 1180. 1005. 970. 855, 835. 755, 730. 720, and 680 cm-l. MS (m/e): 193 (M+l. 7). 192 (M. 33):193 (12). 178 (12). 177 (64). 163 (14). 152 (10). 151 (65). 137 (8). 135 (13). 133 (8). 123 (8). 111 (8). 110 (23). 109 (16). 95 (10). 83 (13). 74 (10). 73 (100). 59 (47). 45 (13). 43 (14). Round** : C, 75.17; H, 11.07. Calc. for C₁₂H₂₀Si : C, 74.92; H, 10.48.

*(E)-1-Trimethylsilyl-4-(1-cyclohexenyl)-1-buten-3-yne (6b) - This compound, which was ob*tained in 81 % yield from (1-cyclohexenyl)ethynylzinc chloride (12g) and (E)/(Z)-11, had: b.p. 77 \textdegree C/0.15 Torr. ¹H NMR (CDCl₃): ∂ 6.38 *(IH, d, J = 19.2 Hz, H1 or H2), 6.15-6.08 <i>(IH, m, H2), 6.06* **(1H. d. J = 19.2 Hz. H2 or Hl). 2.20-2.16 (4H. br m. H3' and H6'). 1.72-1.50 (4H. br m. H4' and HS'), and 0.08 ppm** (9H; 8. **Si(CH3)3).**

MS (m/e) : 205 (M+l. 10). 204 (M. 15). 190 (13). 189 (71). 164 (16). 163 (100). 150 (IO). 145 (12). 126 (9). 123 (23). 109 (8). 83 (12). 73 (23). 72 (8). 59 (28). 45 (8). Found : C. 76.32; H. 9.95. Calc. for C13H20Si : C. 76.39; H. 9.86.

Synthesis of (E)-1-trimethylsilyl-1-octen-3-yne (5e) from 1-hexynyltrimethylstannane $(12e)$ and $(E)/(Z)$ -11 - A solution of Pd(PPh₃)₄ (2.5 g, 2.2 mmol) and $(E)/(Z)$ -2-(bromovinyl)trimethylsilane (11) (7.5 ml, 49 mmol) (E/Z = 93.5/6.5) in DMF (100 ml) was cooled at 0 °C. 1-Hexynyltrimethylstannane ($12e$) (10.7 g, 43.5 mmol) was added and the resulting mixture was stirred for 0.5 h at 0 ^oC, then maintained at 80 ^oC for 1.5 h. During this time ca. 95 % of (E)-11 was consumed. The mixture **was poured into ice-water and extracted with pentane (3 x 100 ml). The combined extracts were fitered, dried** and concentrated in vacuo. The residue was purified by MPLC on a silica gel column, followed by fractional distillation. (E)-1-Trimethylsilyl-1-octen-3-yne (5e), having stereoisomeric purity higher than 99 %, was obtained in 53 % yield. B.p. 100-101 °C/25 Torr. ¹H NMR (CDCl₃): ∂ 6.18 (1H, d, J = 19.2 Hz, **Hl). 5.82 (1H. dt, J = 19.2 and 2.0 Hz, H2). 2.27 (2H. td, J = 6.6 and 2.0 Hz. H3 and HJ). 1.61-1.31 (4H. br m. H6 and H7), 0.94 (3H. t, J = 7.1 Hz, H8), and 0.08 ppm (9H, s, Si(CH₃)₃). MS (m/e): 180 (M, 9). 165 (81). 139 (21). 138 (31). 123 (44). 109 (16). 97 (16). % (16). 95 (18). 85 (13). 83 (32). 73 (58). 60** (9), 59 (100), 45 (33), 43 (38). Found : C, 73.05; H, 11.12. Calc. for C₁₁H₂₀Si : C, 73.25; H, 11.14.

(E)-6-Trimethylsilyl-2-methyl-S-hexen-3-yn-2-o1 (17) - **Deareated 2-methyl-3-butyn-2-01 (Z6)** $(12.0 \text{ g}, 143 \text{ mmol})$ and 2-(bromovinyl)trimethylsilane (11) $(28.9 \text{ g}, 161 \text{ mmol})$ (E/Z = 93.3/6.7) were added to a mixture of BnEt₃N⁺Cl⁻ (1.69 g, 5 mmol), CuI (1.63 g, 8.5 mmol), and Pd(PPh₃)₄ (4.96 g, 4.3 **mmol) in benzene (128 ml). A deareated 10 % aqueous solution of NaOH (268 ml) was added and the** resulting mixture was stirred at 20 °C for 4 h. After this time the mixture was treated with a large excess of a saturated NH₄Cl solution, and extracted with pentane (4 x 100 ml). The extracts were washed with a saturated NH_ACl solution, and water until neutrality, filtered through Celite, dried and concentrated *in vacuo*. *The* **residue was purified by MPLC on a silica gel column, using a mixture of hexane and ether (9/l ratio) as eluant. The collected fractions were concentrated, diluted with pentane, and filtered through Celite. The filtrate was concentrated and fractionally distilled to give compound 17 (21.8 g. 83.8 % yield): b.p. 108- 109 "C/13 Torr. lH NMR (CC4): a 6.26 (1H. d. J = 19.3 Hz. H6). 5.88 (1H. d. J = 19.3 Hz. HS). 2.89** (1H, s, OH), 1.47 (6H, s, H₁ and CH₃), 0.08 ppm (9H, s, Si(CH₃)₃). MS (m/e): 182 (M, 46), 181 (11), **167 (28). 101 (9). 91 (11). 76 (11). 75 (100). 73 (19). 59 (25). 43 (14). IR (film): 3350. 2980. 2960. 2900, 1575, 1460. 1440. 1405, 1375, 1365. 1330, 1310. 1250. 1210. 1165. 1140. 1070. 980. 950. 860.** 840, 740, 725, 690, and 660 cm⁻¹. Found : C, 65.73, H, 9.86. Calc. for C₁₀H₁₈OSi : C, 65.87; H, 9.95.

(E)-2 -TrimethylsiIyl-1 -buten-3-yne (5f) - **A solution of compound 17 (21.3 g. 115 mmol) in tetrabydronaphtalene (60 ml) was reacted under a nitrogen atmosphere with solid NaOH (4.69 g. 117 mmol).** The mixture was stirred at 125 °C for 1 h, cooled and subsequently distilled at 13 Torr, while the temperature was gradually increased to 100 °C. The crude reaction product, which was collected in a trap maintained at -78 °C, was washed twice with a saturated NaCl and Na₂CO₃ solution, dried, and distilled to give **compound** *Sf* **(9.84 g, 67.7 % yield). GLC analysis showed that compound** *Sf* **had 97 % chemical purity**

and stereoisomeric purity higher than 99 %. B.p. 120 °C. ¹H NMR (CCl₄): ∂ 6.45 (1H, d, J = 19.3 Hz, **H1)**, 5.85 (1H, dd, $J = 19.3$ and 2.1 Hz, H2), 2.80 (1H, d, $J = 2.1$ Hz, H4), 0.10 ppm (9H, s, Si(CH₃)₃). **MS (m/e): 124** (M. **8). 111 (13). 109 (100). 83 (35). 73 (13). 67 (13). 59 (54). 53 (24). 45 (16). 43 (44). IR (film): 3300. 3ooO. 2980. 2900. 2100. 1300. 1260. 1250. 1210. 1050. 975, 865. 840. 770. 720. 690. and 610 cm-l. Owing to its volatility, a** satisfactory **elemental analysis of Sf could not be obtained.**

(lE,3Z)-1-Trimethylsilyl-l,3-nonadiene (7a) - In **a typical procedure (Table 3, entry 3). 2-methyl-2-butene (1.49 g, 21.3 mmol) was added during 20 min to a 10 M solution of borane-dimethylsulfide** complex (1.03 ml, 10.3 mmol) in THF (11.7 ml), cooled at -20 °C. The resulting solution was stirred at 20 **OC for 2h and. subsequently, added during 10 min to a solution of (E)-1-trimethylsilyl-1-nonen-3-yne (Sa)** (2.0 g, 10.3 mmol) in THF (2.1 ml), cooled at -10 °C. The mixture was stirred at 20 °C for 3 h and, then, **concentrated at 20 Torr in order to eliminate dimethylsulfide. The resulting solution was diluted with THF** (5 ml) and cooled at 0° C. Acetic acid (6.4 g, 86 mmol) was added and the mixture was stirred at 60 $^{\circ}$ C for 7 **h. After cooling at 0 "C. 3 N NaOH (38 ml) was slowly added. Finally. 35 % hydrogen peroxide (2.5 ml) was slowly added, while the temperature was maintained lower than 40 "C. The resulting reaction mixture** was stirred at 40 °C for 3 h, then cooled to 20 °C, diluted with a saturated NaCl solution and extracted with **hexane (4 x 50 ml). The combined extracts were washed with water, dried. concentrated** *in wacuo.* **and purified by MPLC on a silica gel column, using hexane as eluant. The collected chromatographic fractions** were concentrated to give compound 7*a* (1.43 g, 71 % yield). ¹H NMR (CDC1₃): δ 6.81 (1H, ddd, J = **18.2. 10.7. and 1.1 Hz. H2). 5.92 (1H. t, J = 10.8 Hz, H3). 5.82 (1H. d, J = 18.2 Hz. Hl). 5.43 (lH, dt, J = 10.8 and 7.6 Hz. H4). 2.26-2.16 (2H. m. H5). 1.45-1.22 (6H. m. H6. H7 and H8). 0.93-0.83 (3H. m. H9).** 0.08 ppm (9H, s, Si(CH₃)₃). Decoupling experiments showed that $J_{(H1-H2)} = 10.8$ Hz, $J_{(H2-H3)} =$ **10.6 Hz. J(H3_H4) = 18.2 Hz. MS (m/e): 196 (M. 5). 181 (17). 126 (5). 125 (11). 112 (6). 111 (17). 97 (6). 85 (6). 74 (8). 73 (100). 59 (67). 55 (48). 52 (45). 45 (11). 42 (10). IR (film): 3020. 2960. 2930. 2880. 2860. 1635, 1575. 1470. 1460, 1400. 1380. 1260. 1250. 1000. 960. 865. 840. 770. 730. 710. 690.** and 615 cm⁻¹. Found : C, 73.66; H, 12.53. Calc. for C₁₂H₂₄Si: C, 73.88; H, 12.32. GLC analysis showed that compound 7a had 99.3 % stereoisomeric purity.

Two other different procedures to prepare 7a, which were based on the hydroalumination or the **hydromagnesiation of Sa. followed by protonolysis (Table 3. entries 1 and 2) gave unsatisfactory results.**

(IE,3Z)-1 -Trimethylsilyl-4-phenyl-1,3-butadiene (7b) - **In a typical experiment (Table 3, entry 5).** neat diisobutylaluminium hydride (0.88 g, 6.16 mmol) was added to a solution of (E)-1-trimethylsilyl-4phenyl-1-buten-3-yne $(5b)$ (1.12 g, 5.6 mmol) in ether (6 ml). The mixture was stirred at 35 °C for 30 h, **then cooled to 20 "C. carefully poured into ice-cold 3 N NaOH (28 ml). and extracted with hexane (4 x 30 ml). The combined extracts were washed with a saturated NaCl solution, dried, and concentrated in vacua to** give compound 7b (0.42 g, 38 % yield): ¹H NMR (CCl₄): ∂ 7.31-7.12 (5H, br m, C₆H₅), 7.00 (1H, dd, J **= 18.3 and 11.1 Hz. H2). 6.37 (1H. d. J = 11.5 Hz, H4). 6.16 (1H. t, J = 11.1 Hz, H3). 5.95 (1H. dd. J = 18.3 and 0.9 Hz. HI). 0.08 ppm (9H. s. Si(CH3)3). MS (m/e): 202 (M. 8). 187 (14). 145 (9). 129 (13).** 128 (47), 73 (46), 59 (100), 58 (11), 45 (11), 43 (29). Found : C, 77.41; H, 9.03. Calc. for C₁₃H₁₈Si : **C. 77.15; H. 8.96. GLC analysis showed that compound** *7b* **had 96.2 % stereoisomeric purity.**

The preparation of compound *76* **by hydroboration of** *Sb,* **according to the procedure followed to prepare compound 70. afforded the desired dienylsilane in 59 96 yield and 91.9 % stereoisomeric purity (Table 3. entry 6).**

Poorer results were obtained by hydroalumination of 5b in hexane solution at 50 °C (Table 3, entry 4).

(1E.3Z)-1,4-Bis(trimethylsilyl)-1,3-butadiene (7~) - **Neat diisobutylaluminium hydride (5.06 g, 35.6 mmol)** was added to a solution of 1,4-bis(trimethylsilyl)-1-buten-3-yne $(5c)$ (6.35 g, 32.4 mmol) in ether (33 ml), which was maintained at 20 °C. The mixture, which was monitored by GLC, was stirred at 35 ^oC for 7 h, then cooled to room temperature, carefully poured into ice-cold 3 N NaOH (162 ml), and ex**tracted with pentane (4 x 50 ml). The combined extracts were washed with a saturated NaCl solution, dried,** and fractionally distilled to give compound 7c (5.6 g, 87 % yield): b.p. 80 °C/21 Torr. ¹H NMR (CCl₄): **a 6.78-6.63 (2H. br m. H2 and H3). 5.83 (1H. br d. J = 16.6 Hz. Hl). 5.61 (lH, d. J = 11.2 Hz. H4). 0.16** and 0.09 ppm (18 H, 2s. Si(CH₃)₃). MS (m/e): 198 (M, 6), 183 (11), 181 (6), 131 (8), 110 (22), 75 (13), **73 (100). 69 (30). IR (film): 3050, 2960, 2900, 1545. 1410, 1260, 1250. 1175, 1045, 1000, %5. 870, 835, 760. 745. 730. 685, 670. and 615 cm-l. Found: C. 59.99; H. 11.32. Calc for CIoH2gSi : C, 60.52; H. 11.17. GLC analysis showed that compound 7c had 99.4 96 stereoisomeric purity (Table 3, entry 7).**

General procedure for the diastereoselective alkenylation of (E)/(Z)-2-(bromovinyl)trimethyisilane (II) - In **a typical experiment, a 1 M hexane solution of diisobutylaluminium hydride** (72.5 ml, 72.5 mmol), was slowly added to a freshly distilled 1-alkyne (72.5 mmol). The mixture was stirred for 6 h at 50 °C and, subsequently, added to a solution of $(E)/(Z)$ -11 (13.8 g, 76.8 mmol) ($E/Z =$ **94.4/5.6)** and $Pd(PPh₃)₄$ (2.50 g, 2.2 mmol) in THF (100 ml), cooled at 0 °C. The mixture was stirred for **16 h at 42 Oc. then poured into ice-water and extracted with ether (4 x 100 ml). The combined extracts were filtered, dried, and concentrated** *in vacua.* **The residue was diluted with pentane, filtered through Celite. concentrated** *in vucuo.* **and purified by MPLC on a silica gel column. using hexane as eluant. Fractional distillation of the collected chromatographic fractions gave the desired (lE,3E)-1-trimethylsilyl-1,3 aikadiene (8) having stercoisomeric purity higher than 99 %.**

Some physical and spectroscopic properties of the (lE.3E)-I-trimethylsilyl-1.3-alkadienes prepared using this procedure. i.e 8a and 8b, are reported below.

(lE\$E)-I-Trimethylsilyl-1,3-nonadiene (8a) - **This compound, whiih was prepared in 74.7 % yield** from 1-heptyne and $(E)/(Z)-11$, had: b.p. 79-80 °C/6 Torr. ¹H NMR $(CCl₄)$: ∂ 6.37 (1H, dd, J = 18.6 and **10.1 Hz, H2), 5.95 (1H, br dd, J = 14.1 and 10.1 Hz, H3), 5.70-5.53 (2H, m, H1 and H4), 2.12-2.00 (2H, m. H5). 1.46- 1.22 (6H, br m. H6, H7. and H8). 0.89 (3H. br t, J = 6.6 Hz, H9), 0.05 ppm (9H. s.** Si(CH₃)₃). MS (m/e): 196 (M, 16), 181 (54), 125 (18), 112 (13), 111 (29), 109 (10), 73 (54), 59 (100), **45 (15). 43 (15). IR (film): 3050. 2960. 2930. 2870, 1640. 1580, 1470, 1460. 1380, 1310, 1260. 1245. 1180. 1000. 950. 860. 830. 770. 720. 685. and 610 cm-l. Found : C. 73.92; H. 11.51. Calc for C12H24Si : C. 74.14; H. 11.40.**

(IE,3E)-1 -Trimethylsilyl-S,S-dimethyl- ,3-hexadiene (8b) - **This compound, which was prepared** in 84 % yield from 3,3-dimethyl-1-propyne and $(E)/(Z)$ -11, had: b.p. 96 °C/29 Torr. Lit^{20a} b.p. 87-89 \textdegree C/136 Torr. ¹H NMR (CDC1₃): ∂ 6.37 (1H, dd, J = 18.3 and 9.5 Hz, H2), 5.90 (1H, dd, J = 15.4 and 9.5 **Hz, H3). 5.67 (1H. d. J = 15.4 Hz. H4). 5.64 (1H. d. J = 18.3 Hz, Hl), 1.04 (9H, s. C(CH3)3), and 0.06 ppm (9H. s. Si(CH3)3). MS (m/e): 182 (M. 8). 167 (11). 107 (6). 74 (8). 73 (100). 59 (32). 57 (14). 45** (8), 43 (8). Found : C, 72.62; H, 12.51. Calc. for C₁₁H₂₂Si : C, 72.44; H, 12.17.

(IE,3E)-1-(2-Thenoyl)-I,3-nonadiene (20) - **A suspension of anydrous AlC13 (1.2 g. 8.9 mmol) in** dichloromethane (45 ml) was stirred for 20 min at room temperature, then it was cooled to 0 °C. 2-Thenoyl chloride (19) (11.3 g, 8.9 mmol) was added and the mixture was stirred at 20 °C for 1 h, then cooled to 0 ^oC. (1E,3E)-1-Trimethylsilyl-1,3-nonadiene (8a) (1.75 g, 9.9 mmol) was added and the mixture was stirred at 0 $^{\circ}$ C for 1h and 45 min, and subsequently at 20 $^{\circ}$ C for 2 h. This mixture was poured into ice-cold water and NaHCO₃ and extracted with hexane $(3 \times 100 \text{ ml})$. The combined extracts, after usual work up, were **concentrated and purified by MPLC on a silica gel column. using a mixture of hexane and ether (95/5) as eluant. The collected chromatographic fractions were concentrated** *in vacua to* **give compound 20 (0.73 g,**

37. % yield). 1H NMR (CCL): a 7.75 (lH, dd. J = 3.8 and 1.1 Hz. H3' or HS'). 7.63 (1H. dd, J = 4.9 and 1.2 Hz, H5' or H3'), 7.44 (1H, dd of m, $J = 14.9$ **and 10.2 Hz, H2), 7.14 (1H, dd,** $J = 4.9$ **and 3.8 Hz, H4'), 6.79 (1H. d, J = 14.9 Hz. Hl). 6.32-6.25 (2H. m. H3 and H4). 2.25-2.16 (2H, m. H5). 1.52-1.22 (6H, br m. H6. H7. and H8), 0.90 ppm (3H, br t, J = 6.9 Hz, H9).**

$$
s \sqrt[4]{s} \sqrt[3]{\frac{1}{2} \frac{1}{2} \frac{1
$$

MS We): 234 (M. 6). 164 (14). 163 (100). 135 (10). 111 (43). 91 (63). 83 (6). 67 (6). 41 (9). 39 (14). Found : C. 72.03; H 8.01. Calc. for C₁₄H₁₈OSi : C, 71.75; H, 7.74. GLC analysis showed that **compound 20 had chemical and stereoisomeric purity higher than 98.5 96.**

 $(ZE,4E)$ -2,4-Decadienal (22) - A solution of TiCl₄ (0.86 ml, 7.8 mmol) in dichloromethane (15 ml) was added during 40 min to a solution of dichloromethyl methyl ether (0.92 g, 7.7 mmol) and (1E,3E)-1trimethylsilyl-1.3-nonadiene (8a) (1.50 g, 7.6 mmol) in dichlorometane (15 ml) cooled at -78 °C. The **resulting mixture was stirred at -78 'C for 1 h and 20 min. then hydrolyxcd by** *addition* **of 50 % aqueous** methanol at -20 °C. The organic layer was separated and the aqueous phase was extracted with dichloromethane (3 x 10 ml). The combined organic extracts were washed with 5 % aqueous NaHCO₃ solution, dried and concentrated. The residue was purified by MPLC on a silica gel column, using a mixture **of hexane and ether (9/l) as eluant. The eluted chromatographic fractions were analyzed by GLC and concentrated in vacuo to give compound 22 (0.47 g, 40.4 % yield):** ¹H NMR (CDC1₃): ∂ 9.54 (1H, d, J = 8.1 Hz, H₁), 7.10 (1H, dd of m, $J = 15.1$ and 10.0 Hz, H₃), 6.34-6.27 (2H, m, H₄ and H₅), 6.07 (1H, dd, $J =$ **15.1 and 8.1 Hz, H2). 2.27-2.17 (2H. m. H6). 1.60-1.20 (6H. m. H7. H8. and H9). 1.00-0.85 ppm (3H. br t. H10).** ¹H NMR (CDCl₃, in the presence of 0.5 mol % of Resolve-Al EuFOD): ∂ 11.20 (1H, d, J = 8.0 **Hz, Hl), 8.36 (1H. dd. J = 14.8 and 8.0 Hz. H2). 7.85 (1H. dd. J = 14.8 and 10.8 Hz. H3). 6.85 (lH, dd, J** $= 14.9$ and 10.8 Hz, H4), 6.50 ppm (1H, dt, J = 14.9 and 6.9 Hz, H5). ¹³C NMR (CDC1₃): ∂ 194.2, 153.0, **147.5. 130.2, 128.8. 32.8. 31.0. 27.9, 22.0. 13.5 ppm. MS (m/e): 152 (M. 56). 123 (37). 109 (37). 108 (16). 95 (8). 83 (11). 81 (100). 67 (18). 55 (13). 41 (39). 39 (30). IR (film): 3360. 3040, 2960. 2930. 2880. 2740. 1690, 1860. 1640. 1600. 1590, 1470. 1460. 1435. 1380. 1290, 1165, 1120. 1110. 1010 cm-*. GLC analysis showed that compound 22 had 97.3 % stereoisomeric purity and chemical purity higher than 99 %.** 2,4-Dinitrophenylhydrazone of 22 had: m.p. 141 °C (from ethanol). Lit.³⁷ m.p. 142-144 °C.

(2E,4Z)-2,4-Decadienal (23) - This compound was prepared in 45.1 % yield from (1E,3Z)-1**trimethylsilyl-1.3-nonadiene (70) and dichloromethyl methyl ether (21). according to the same procedure employed to prepare compound 22.**

The spectroscopic properties of compound 23. which had stereoisomeric purity higher than 98.6 %. matched those previously reported 38.

 (E) -5-(2-Thienyl)-2-penten-4-yn-1-ol (27) - A deareated solution of 2-ethynylthiophene (25)²⁹ (6.5g, 60 mmol) and (E)-3-iodo-2-propen-1-ol $(26)^{18c}$ (11.0 g, 60 mmol) in benzene (75 ml) was added to **a** stirred mixture of Pd(PPh₃)₄ (2.6 g, 2.3 mmol), CuI (0.87 g, 4.6 mmol) and triethylamine (14 ml) in **benzene (85 ml). The resulting mixture was stirred at room temperature for 2 h. then diluted with ether and** washed repeatedly with a saturated NH₄Cl solution. The organic phase was washed with water, dried, and concentrated *in vacuo*. The residue was purified by MPLC on a silica gel column, using a mixture of **hexane and benzene (2/l) as eluant. The collected chromatographic fractions were fractionally distilled to give compound 27 (5.3 g. 54.3 96 yield): b.p. 115-116 Y!/O.O3 Torr. tH NMR (CDC13): a 7.48-7.22 (2H. m. H3' and HS'). 7.08 (1H. dd. J = 5.0 and 3.6 Hz. H4'). 6.52 (1H. dt. J = 15.3 and 4.5 Hz. H2). 6.10 (1H. d. J - 15.3 Hz. H3). 4.41 (2H. br d. J= 4.5 Hz, Hl). and 2.00 ppm (1H.** br 8, OH).

JR (film): 3940. 3820. 3340. 3100. 3030. 2910. 2860. 2700, 2190. 1990. 1940. 1900. 1800, 1755, 1725, 1660. 1630. 1600. 1530, 1445. 1425. 1370. 1290. 1270. 1240. 1195. 1135. 1090. 1040, 975, 945, 900, 845. 825. and 695 cm-l. MS (m/e): 165 (M+l. 13). 164 (M. 100). 163 (28). 136 (20). 135 (87). 134 (20). 121 (27). 120 (16). 108 (40). 91 (51). 89 (13). 77 (13). 69 (14). 63 (15). 43 (20). Found : C. 66.oo; H, 5.01. Calc. for CgHgOS : C. 65.82; H. 4.91. GLC analysis showed that compound **27 had chemical and stereoisomeric purity higher than 99 %.**

(E)-S-(2-Thienyl)-2-penren-4-ynal (24) - A **mixture of compound 27 (5.0 g. 30 mmol) and** manganese dioxide³⁰ (100g, 1.15 mol) in ether (375 ml) was stirred for 2 h at room temperature. The **mixture was filtered** through **Celite and the solid obtained was washed repeatedly with ether (4 x 50 ml). The** collected organic phases were concentrated and the residue was crystallyzed from hexane to give compound **II** (2.6 g, 53.7 % yield): m.p. 40-41 °C. Lit²⁷ m.p. 41.5 °C. ¹H NMR (CCl₄): ∂ 9.56 (1H. d. J = 7.3 **Hz, H1). 7.38 (1H, d, J = 5.0 Hz, H5'). 7.29 (1H, d, J = 3.6 Hz, H3'). 7.02 (1H, dd, J = 5.0 and 3.6 Hz, H4')**, 6.74 (1H, d, J = 15.9 Hz, H3), and 6.47 ppm (1H, dd, J = 15.9 and 7.3 Hz, H2).

MS (m/e): 163 (M+l. 10). 162 (M. 82). 134 (93). 133 (15). 108 (100). 93 (10). 90 (19). 89 (38). 87 (lo), 75 (15). 74 (18). 69 (24). 63 (28). 62 (14). 58 (14). 50 (13). 45 (23). IR (film): 3300. 3100. 3005. 2980, 2820, 2725, 2225, 2180. 1600. 1580, 1420. 1390, 1355. 1285, 1255. 1220. 1200, 1155. 1080, 1045, 950, 845, 830, and 700 cm-l.

3*-Methyl-4,5,6,7-tetrahydroindan-1-one (28a) and 3-methyl-5,6,7,7a-tetrahydroindan-1*one (29) - (E)-1-Trimethylsilyl-4-(1-cyclohexenyl)-1-buten-3-yne (6b) (6.4 g, 31.3 mmol) was added to **a mixture of concentrated sulfuric acid (35.5 ml) and acetic acid (177 ml). The nsulting solution was heated** at 60 °C for 40 min, then cooled to room temperature, poured into ice-cold water, and extracted repeatedly with hexane $(5 \times 100 \text{ ml})$. The collected organic extracts were washed with water, saturated NaHCO₃ **solution, and water, dried and concentrated. GLC/MS analysis of the residue showed the presence of two** isomeric compounds in a ca. 22/78 ratio. This residue was purified by MPLC on a silica gel column, using **a mixture of hexane and ether (8/2) as eluant. Fractional distillation of the Fist eluted fractions gave 3 methyl-5,6,7,7a-tetrahydroindan-l-one (29) (1.06 g. 17.5 96 yield). The structure of this ketone was established on the basis of a 2D lH NMR correlation (COSY) spectrum. Compound 29 had: b.p. 105** $C(7.5$ Torr. ¹H NMR (CDCl₃): ∂ 7.98-7.92 (1H, m, H4), 2.89 (1H, dd, J = 17.5 and 7.0 Hz, 1H2), 2.82-2.68 (1H, m, 1H5), 2.82-2.58 (3H, m, 1H5 and 1H6), 2.35-2.25 (2H, m, 1H2 and H7a), 2.10-1.92 (1H, m, **H3). 1.92-1.75 (1H. m. lH7). 1.42 (3H. d, J - 6.0 Hz. CH3). 1.30-1.15 ppm (1H. m. lH7).**

JR (Elm): 3430. 3030. 2960. 2930. 2870. 2820. 1720. 1660, 1450. 1430. 1380, 1340. 1330. 1300, 1270, 12SO. 1225. 1210. 1120. 1070. 1055. 1005. 970. 940. 920. 900. 875. 855. 785. 745. 655, and 620 cm-t. M8 (m/e): ISO (M. 24). 135 (14). 108 (100). 107 (12). 93 (21). 91 (10). 80 (35). 79 (75). 77 (12). 52 (IO),

41 (13). 39 (14). Found : C. 74.85: H. 9.51. Calc. for C10H140 : C. **74.%; H. 9.39.**

Fractional distillation of the less easily eluted fractions gave 3-methyl-4,5,6,7-tetrahydroindan-1-one $(28a)$ (3.9 g, 65 % yield): b.p. 109 °C/7.5 Torr. Lit³³ b.p. 83-85 °C/1 Torr. ¹H NMR (CDCl₃): ∂ 2.79-**2.69 (lH, m. H3). 2.62 (1H. dd. J = 18.0 and 6.0 Hz. lH2). 2.30 (1H. dt. J = 19.0 and 5.0 Hz. lH4 or lH7). 2.18 (1H. dt, J = 19.0 and 3.0 Hz, lH4 or lH7). 2.15-2.08 (2I-I. m. H7** or **H4). 1.95 (1H. d. J = 18.0** Hz , 1H2), 1.83-1.54 (4H, m, H5 and H6), 1.15 ppm (3H, d, J = 6.0 Hz, CH₃)

IR (film): 3380, 2930. 2870. 2820, 1700, 1650. 1440, 1420. 1395. 1375, 1350. 1295. 1280, 1250. 1235, 1125, 1095. 1060. 1045, 1010. 945. 810. 800, 775, 745. and 720 cm-l. MS (m/e): 151 (M+l. 10). 150 (M. 84). 149 (15). 135 (100). 122 (26). 108 (18). 107 (51). 94 (19). 93 (24). 91 (18). 80 (12). 79 (57). 77 (20). 53 (10). 41 (14). 39 (15).

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